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Transmissivity of Carbon Monoxide in the 2.3 µm Band Region

Final Report

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High Altitude Engineering Laborator Departments of Aerospace Engineering Atmospheric and Oceanic Science

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THE UNIVERSITY OF MICHIGAN

COLLEGE OF ENGINEERING

High Altitude Engineering Laboratory

Departments of

Aerospace Engineering

Atmospheric and Oceanic Science

Final Report

TRANSMISSIVITY OF CARBON MONOXIDE

IN THE

2.3µm BAND REGION

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ORA Project 010808

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α<u>+</u>πγ4.

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ABSTRACT

Line strengths and self and nitrogen broadened half-widths have been determined from high resolution (0.12 cm $^{-1}$) spectroscopic measurements of selected lines in the 2.3 μ m band region of CO. The CO 0-2 total band strength is estimated to be 2.086 \pm 0.146 cm $^{-1}$ (atm-cm) $^{-1}$ STP which is higher than most previously reported values. The line half-widths are also generally higher than those in the literature.

TRANSMISSIVITY OF CARBON MONOXIDE IN THE 2.3µm BAND REGION

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1. INTRODUCTION

This report presents the experimentally obtained results of a study made of the 0-2 absorption spectrum in the 2.3 m band region of carbon monoxide. The line parameters: line strength, and self and nitrogen broadened half widths are given for various principle lines in this spectral region. These results along with a calculated band strength value of 2.086 ± 0.146 cm⁻¹ (atm cm)⁻¹ STP are comparatively higher than many of the previously reported results of experimental and theoretical studies of the first overtone band of CO. Our data analysis technique, which does include wing and base corrections applied to the measured equivalent widths, has produced results which are comparatively higher not only for this present study but also for an earlier study of the fundemental band of CO by Drayson et.al. (1973). In this report we discuss possible reasons for the differences between the results of our study and the results of other authors. brief descriptions are given of the experimental and analytical methods used in this study.

2. EXPERIMENTAL

A complete description of the instrumentation and measurement procedure used in this study can be found in the

report by Chaney (1972). A very brief description is as follows. A Jarrell Ash 1.83 meter Fastie Ebert vaccuum scanning spectrometer was used to make the actual spectroscopic measurements. The source of the infrared radiation was a General Electric 600 watt tungsten - iodide lamp cooled to -78 $^{\circ}$ C with dry ice in ethyl alcohol. The gas sample cell was made from the entrance optics by means of having a sapphire window placed between the entrance optics flange and the spectrometer main frame. The cell dimensions were one meter from the input window to a diagonal mirror, one meter from the diagonal mirror to a spherical mirror, and one meter back from the spherical mirror to the entrance slit. Thus, the total optical path length was 3.0 meters. The detector output signal after passing through a pre-amplifier and phase lock amplifier was a d.c. signal which varied between 0 and +5 volts. This signal was recorded on an analog chart recorder and was also digitized. The data was punched on IBM cards. The data scan rate was about 110 seconds per wavenumber. A signal to noise ratio of approximately 200:1 was obtained for most of the data.

Matheson Company carbon monoxide with a stated purity of 99.5% or higher was used in this study. Nitrogen gas from a large liquid nitrogen storage dewar was used in certian measurements to broaden the CO. The CO and N_2 mixtures were allowed to mix for periods on the order of eighteen

hours before measurements were taken.

For all data scans the gas sample pressures were measured to within 0.01 mm of Hg and the room temperature was kept constant to within $\pm 0.5^{\circ}$ C. The gas sample cell temperature for all measurements was 25° C \pm 2° C.

The 2.3µm band was studied with a Bausch and Lomb grating blazed at 4.0µm and filtered for the second order. The best measured resolution for this grating was 0.12 cm $^{-1}$. The band was isolated with a 2.07µm-2.708µm band pass optical coating laboratory interference filter. Eight regions of the band were investigated (Table 1). These regions were chosen because of the relative minimum of line overlap.

Table 1
Regions of the band studied

| Region | Wavenumber [cm ⁻¹] | C ¹² O ¹⁶ Lines |
|--------|--------------------------------|---------------------------------------|
| 1 | 4152-4156 | P23 |
| 2 | 4193-4197 | P15 |
| 3 | 4230-4232 | P7 |
| 4 | 4233-4246 | P4-P6 |
| 5 | 4255-4276 | R3-P1 |
| 6 | 4283-4287 | R6 |
| 7 | 4310-4315 | R15-R16 |
| 8 | 4332-4336 | R24-R25 |

3. ANALYSIS

The analysis method closely followed in this study was used by Drayson et.al. (1973) in the experimental study of the

transmissivity of CO in the 4.6 µm region. A complete description of the method can be found in that report.

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Basically, the method of curve of growth was used to determine the values of the line strenths and line half widths of those spectral lines investigated. Measurements using the trapezoidal rule were made of the equivalent widths for each spectral line in which adequate data was available. For those lines in which weak isotopic line blending was observed, a correction equal to the product of the isotope line strength and absorber path length was subtracted from the total equivalent width. An iterative procedure was used to make the line parameter calculations. Initial values of the line parameters were taken from Kunde (1967). Additional corrections were made to the measured equivalent widths. Wing and base corrections from Korbe et.al. (1968) were applied. The wing and base corrected equivalent widths were corrected for doppler broadening. Correction expressions for a Voigt profile line were obtained from Yamada (1968).

The line strength and half width values calculated during an iteration were averaged and used as improved estimates for the following iteration. The iterative procedure continued until the difference between the newly calculated values and the previously calculated values differed by no more than 0.0002.

For the six spectral lines in which enough useable data

was available to calculate all three line parameters the Herman Wallis (1955) formula was used to calculate the band strength of the first overtone of ${\rm C}^{12}{\rm O}^{16}$. An average of the six calculated band strengths was then used in the Herman Wallis formula to compute the line strengths of the other spectral lines in the study. These line strength values were then used to compute half-width values for those lines having enough self and/or N₂ broadened data.

Michel

4. RESULTS AND DISCUSSION

The line strengths and line half-widths of the R6, R3, R2, R1, R0 and P15 lines determined in this study are presented in Table 4. A band strength for the first overtone of C¹²O¹⁶ was calculated from the line strength of these six lines. These results are given in Table 2. For each spectral line the individual measurements are assumed to be normally distributed about the average line strength. One standard

Table 2
Band strength from individual measurements

| Line | Number of Measure- ments | Line strength cm ⁻¹ (atm.cm) ⁻¹ | Band Strength cm ⁻¹ (atm.cm) ⁻¹ STP | Standard Deviation |
|------|--------------------------------|---|---|-----------------------|
| R6 | 1.0 | 0.0876 | 1,9239 | +3.6 |
| R3 | 10 | 0.0678 | 2.0053 | 4 .9 |
| R2 | 8 | 0.0562 | 2.1193 | 4 4.4 |
| Rl. | 8 | 0.0358 | 1.9531 | 7 3.2 |
| R0 | 10 | 0.0207 | 2.2191 | $\overline{\pm}6.2$ |
| P15 | 6 | 0.0292 | 2.1248 | $\frac{1}{2}$.0 |

deviation, stated in percentage of the average, is calculated from the individual measurements for each of the six lines.

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The average band strength from these six measurements is 2.057 cm⁻¹(atm.cm)⁻¹STP. By assuming that the CO gas samples had an isotopic composition clos to that of atmospheric CO. the average band strengths of the first overtones of $c^{13}o^{16}$, $c^{12}o^{18}$, and $c^{12}o^{17}$ were calculated to be 0.023, 0.004, and 0.0007 cm⁻¹ (atm.cm) -1 STP, respectively. Thus, the total band strength of the first overtone of CO is estimated to be 2.086 + 0.146 cm⁻¹ (atm.cm) -1STP. A discussion of the error estimate follows later in this section. This band strength is approximately 9% higher than the average of 1.88 cm⁻¹ (atm. cm) $^{-1}$ STP obtained from the previous studies of the 2.3 m band region listed in Table 3. The larger value of our band strength is due primarly to the wing and base corrections of the equivalent widths in our calculations. Our band strength value does compare most favorably with the value obtained by Schurin and Ellis (1966) who used the Wilson-Wells-Penner-Weber self broadening technique in their theoretical study.

The self-broadened line widths determined from this study are compared in Figure 1 with results from recent studies. Out values are approximately 10% greater than those obtained by Kostowski and Pass (1961) in their direct measurement study of the first overtone band of CO. There is fairly good agreement between our nitrogen broadened half-

Table 3

Band Intensities of the 2.3_{um} band of CO

Mary.

| year | Author | Band Intensity cm ⁻¹ (atm.cm) ⁻¹ STI | |
|------|------------------------|---|--|
| 1951 | Penner and Weber | 1.83 | |
| 1954 | Vincent and Geisse | 1.83 | |
| 1961 | Kostkowski and Bass | 1.95 | |
| 1962 | Oppenheim and Goldring | 1.66 | |
| 1965 | Armstrong and Welsh | 1.77+0.05 | |
| 1966 | Schurin and Ellis | 2.15+0.1 | |

width values and those values from recent studies. Although, and our half-width values are higher than those of other authors. The nitrogen broadened half-widths results are compared in Figure 2 with the results from recent investigations.

A discussion of the errors and uncertainties involved with this study is useful when considering the validity of our results. Reichle (1969) has extensively discussed the inherent difficulties in making accurate determination of the line parameters in a study of this type. Possibly the greatest margin of error in this study lies in the difficulty in accurately locating the baselines of the absorption line. For those lines in which there were few or no closely neighboring isotopic lines we took an average of as many as 30 data points, (i.e. each data point corresponds to approximately 0.008 cm⁻¹) on either side of the spectral line.

TABLE 4
Measured Line Intensities and Half-Widths

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| Line | <u>Isotope</u> * | Line Center | S cm ⁻¹ [atm ⁻¹ cm ⁻¹] STP | α _{CO} atm ⁻¹ cm ⁻¹ | α_{N_2} |
|------|------------------|-------------|--|---|----------------|
| R25 | 26 | 4335.02 | .0014 | - | _ |
| R24 | 26 | 4333.04 | .0022 | | - |
| R15 | 26 | 4311.95 | .0364 | | _ |
| R6 | 26 | 4285.00 | .0873 | .0804 | .0674 |
| R3 | 26 | 4274.74 | .0678 | .0849 | .0775 |
| R2 | 26 | 4271.17 | .0562 | .0940 | .0878 |
| Rl | 26 | 4267.54 | .0358 | .0860 | .0806 |
| R0 | 26 | 4263.83 | .0207 | .0857 | .0822 |
| Pl | 26 | 4256.21 | .0188 | .0805 | - |
| P4 | 26 | 4244.26 | .0610 | .0877 | - |
| P5 | 26 | 4240.13 | .0687 | - | - |
| P7 | 26 | 4231.68 | .0747 | - | .0667 |
| P15 | 26 | 4195.11 | .0292 | .0742 | .0567 |
| R7 | 36 | 4193.86 | .0011 | | - |
| Р3 | 36 | 4155.60 | .0006 | - | - |
| P23 | 26 | 4154.19 | .0024 | .0552 | - |
| | | | | | |

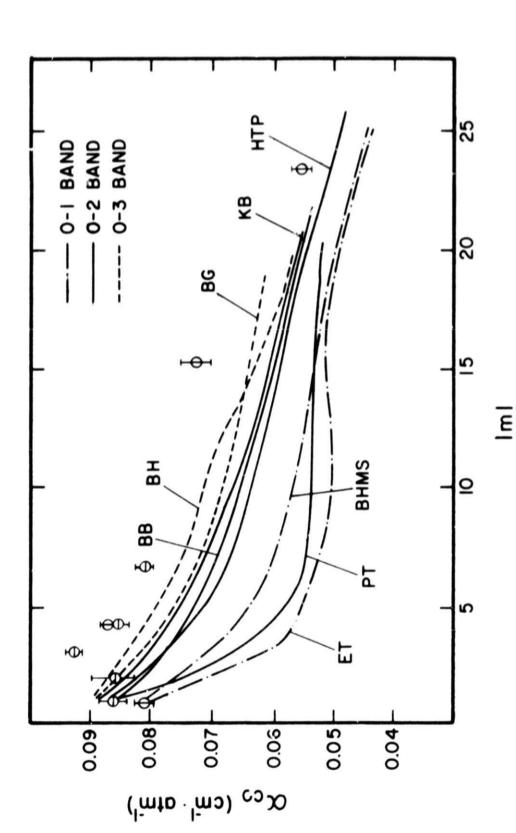
^{*}The notation used to identify the isotope is: $26=C^{12}O^{16}$, $36=C^{13}O^{16}$.

Yet, for those lines having close neighboring isotope lines it was often necessary to take an average of as few as 10 data points for the base transmission line. In addition to the low amounts of noise present in the data we also had shifts in the background levels that were on the order of 1% per hour of data acquisition. This systematic slow change in background was apparently due to self-heating taking place in some part of the instrument. For certain data scans there was also some aperiodic variation in the background level over intervals of a few tenths of a wavenumber. variations were normally quite small in amplitude, yet they were often easily recognizable in the computer drawn plots. The amplitudes of these variations normally were no larger than 1% of the amplitude of the spectral line. The cause of these variations is not fully understood. We feel that errors in measurements of the pressures, the total optical path length of the spectrometer, the temperatures, and our estimate of the isotopic composition of CO were fairly small, about on the order of less than 1%. We estimate the error due to wavenumber calibration and wavenumber shift caused by temperature changes to be less than 1%. Errors in the data transmission system are also very small having magnitudes less than 0.5%.

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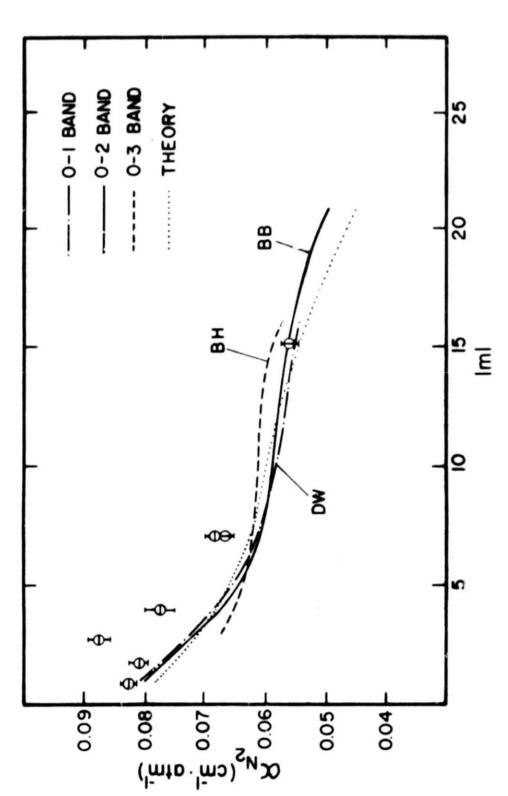
We, thus, estimate that the equivalent widths were measured to within 4% accuracy for most of the data. Due to

the inherent difficulties in accurately measuring the relatively low equivalent widths of much of our data we estimate that our band strength is accurate to within $\pm 7\%$. The error bars in Figures 1 and 2 are estimates of the dispersion of the experimentally obtained half-width values of this study about the plotted averages. Each error bar represents one standard deviation. It is important to note that these error bars do not totally represent the absolute error in the half-width measurements. Since there is some nonlinear dependence of the calculated half-widths upon the individual line strength estimates, the actual error estimates associated with these values may be somewhat greater in magnitude. Figure 3 shows a computer drawn plot of a scan of a portion of the 2.3 μ m absorption spectra of CO.



(1963)-PT; Hunt et.al. (1968)-HTP; Bouanich and Brodback (1974)-BB; 0-3 band of Burch and Gryvnak (1967)-BG; Bouanich and Haeusler (1972)-BH. BHMS; 0-2 band of Kostkowski and Bass (1961)-KB; Plyler and Thibault the 0-1 band of Eaton and Thompson (1959)-ET; Benedict et.al. (1962) Self-Broadened half-widths of CO lines versus | m | results of this work (O-P branch and O-R branch) are shown along with results in Figure 1:

OFGW2D



results of Bouanich and Haeusler and Williams (1968) (0-1 band)-Nitrogen Broadened half-widths of CO lines versus |m | results of this are shown with the theoretical (0-2 band) -BB. (1972) - BH; and results of Draegert DW; and Bouanich and Brodbeck (1974) work (0 - P branch and 0 - R branch) Figure 2:

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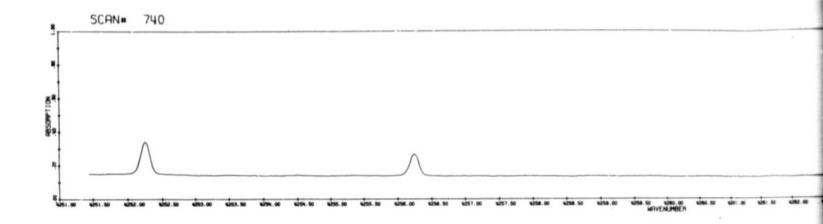
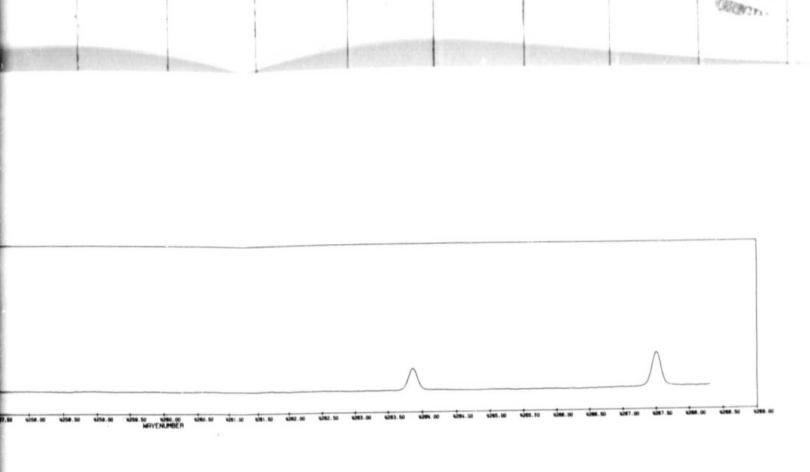


Figure 3: Example of partial spectrum of the 2.3µm CO across the center of the band. Cell length = 300 cm, partial pressure of torr, partial pressure of N₂ = 75.0 torm 25.0°C.

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le of partial spectrum of the 2.3 μ m band of ross the center of the band. length = 300 cm, partial pressure of CO = 5.0 partial pressure of N₂ = 75.0 torr, temperature C.

APPENDIX: DATA SUMMARY

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| Scan No. | Wave Number | Lines CO ¹² O ¹⁶ | Pressure mm Hg | |
|----------|------------------|---|------------------------|----------------|
| | cm ⁻¹ | CO. O. | со | N ₂ |
| 615-616 | 4255-4276 | RO, P1 | 2.5,5 | |
| 618-623 | 4255-4276 | R2, P1 | 5,20,40,80, 160,320 | |
| 624 | 4255-4276 | R0 | 5 | 5 |
| 627 | 4255-4276 | R0 | 5 | 75 |
| 630 | 4255-4276 | RO, P1 | 5 | |
| 636 | 4255-4276 | R2,R1,P1 | 2.5 | |
| 644 | 4255-4276 | P1 | 5 | |
| 659 | 4283-4287 | R6 | 5 | 35 |
| 660 | 4255-4276 | R3, R1 | 5 | 35 |
| 662 | 4230-4232 | P7 | 5 | 35 |
| 667 | 4283-4287 | R6 | 5 | 155 |
| 668 | 4255-4276 | R3, R1 | 5 | 155 |
| 670 | 4230-4232 | P7 | 5 | 35 |
| 671 | 4193-4197 | P15 | 5 | 155 |
| 675 | 4283-4287 | R6 | 5 | 690.7 |
| 676 | 4255-4276 | R3,R2,R1 | 5 | 690.7 |
| 678 | 4230-4232 | P7 | 5 | 690.7 |
| 679 | 4193-4197 | P15 | 5 | 636.5 |
| 683 | 4283-4287 | R6 | 12.5 | |
| 684 | 4255-4276 | R3,R2,R1 | 12.5 | |

APPENDIX: DATA SUMMARY (concluded)

| Scan No. | Wave Number cm-1 | Lines CO ¹² O ¹⁶ | Pressure mm Hg | |
|----------|------------------------|---|---------------------|----------------|
| | | | со | N ₂ |
| 685 | 4233-4256 | P4 | 12.5 | |
| 688 | 4152-4156 | P23 | 12.7 | |
| 691 | 4283-4287 | R6 | 25 | |
| 692 | 4255-4276 | R3,R1,P1 | 25 | |
| 695 | 4293-4197 | P15 | 25 | |
| 699 | 4283-4287 | R6 | 75 | |
| 700 | 4255-4276 | R3,R2,R1 | 75 | |
| 701 | 4233-4256 | P4 | 75 | |
| 703 | 4193-4197 | P15 | 75 | |
| 704 | 4152-4156 | P23 | 75 | |
| 707 | 4283-4287 | R6 | 150 | |
| 708 | 4255-4276 | R3, R2 | 150 | |
| 711 | 4193-4197 | P15 | 150 | |
| 731 | 4283-4287 | R6 | 5 | 15 |
| 732 | 4255-4276 | R3, R2 | 5 | 15 |
| 739 | 4283-4287 | R6 | 5 | 75 |
| 740 | 4255-4276 | R3, R2 | 5 | 75 |
| 742 | 4230-4232 | P7 | 5 | 75 |
| 747 | 4283-4287 | R6 | 5 | 315 |
| 748 | 4255-4276 | R3,R2,R1 | 5 | 315 |
| 751 | 4193-4197 | P15 | 5 | 315 |

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